

Reply to Comment on “Formation of abnormally high density H₂S fluid in sedimentary basins” by Liu *et al.*, 2025

Y. Liu¹, X. Wang^{1,2*}, Y. Song³, H. Shi^{4,5}, I-M. Chou⁶, Q. Wan¹, C. Yu⁷, C. Zhou⁴



Reply to Comment

<https://doi.org/10.7185/geochemlet.2529>

Received 2 July 2025 | Accepted 22 July 2025 | Published 26 August 2025

We acknowledge, with appreciation, the comments from Cui (2025) and his effort in evaluating our recent publication on Raman spectroscopic and microthermometric identification of high density, high purity H₂S fluid inclusions in sedimentary basins (Liu *et al.*, 2025). We are also grateful to the editor for the opportunity to clarify aspects of our manuscript requiring further explanation. Below we respond to the questions raised by Cui (2025).

1. Mississippi Valley-Type (MVT) Pb-Zn Ore Deposits. In Liu *et al.* (2025), we pointed out the major goal of this work as “...to test whether high density H₂S fluids can be formed in sedimentary basins and to evaluate the validity of the proposed mineralising mechanisms for MVT deposits”. Our fluid inclusion observations provide direct evidence confirming the potential for high density H₂S fluid formation. Therefore, in the discussion, we emphasised that such a finding provides critical implications for research on the mineralisation models of MVT Pb-Zn deposits. We acknowledge the point raised by Cui (2025) that sections like Caojunba do not host MVT Pb-Zn mineralisation, which was not the primary focus of Liu *et al.* (2025). The formation of ore deposits is the result of convergence of multiple factors. For the fluid mixing model, formation of an H₂S reservoir is only a prerequisite. In the “Samples and Methods” section, we stated: “The nodular pore-filling quartz-calcite cements are common in the upper Z₁d and the lower to middle Z₂dn dolomites hosting MVT deposits (e.g., Yangjiaping, Zhongling, Caojunba, Nanbeizhen, and Zouma sections)”. This statement plausibly led to the misunderstanding by Cui (2025). Our intended meaning was that the upper Z₁d and lower to middle Z₂dn dolomite represent host rocks for MVT Pb-Zn deposits, and the aforementioned sections are typical Z₁d-Z₂dn outcrops. In fact, several large MVT Pb-Zn deposits, such as Bingdongshan, Dongjiahe, and Muyuhe, have been discovered within Z₁d dolomite in South China (see Supplementary Information in Liu *et al.*, 2025).

2. Calcite Purification Model. Cui (2025) argues that H₂S cannot coexist with calcite, as high H₂S concentrations indicate strongly acidic fluids, which would lead to calcite dissolution.

Consequently, he dismisses the proposed calcite purification model—where calcite precipitation consumes CO₂ and enriches H₂S—as unfeasible. However, this interpretation overlooks well documented evidence that TSR-derived CO₂ can be sequestered by secondary carbonate minerals (Worden *et al.*, 1996; Bildstein *et al.*, 2001; Zhu *et al.*, 2005). Notably, the negative carbon isotope values of such carbonates (e.g., <−5 ‰, VPDB) serve as a key indicator of TSR involvement (Krouse *et al.*, 1988; Jiang *et al.*, 2014; Jenden *et al.*, 2015). For example, in Sichuan (South China) and Southern Permian basins (Poland), TSR significantly alters hydrocarbon compositions, with H₂S concentrations exceeding those of CO₂, while carbonate cements are common (Liu *et al.*, 2014; Li *et al.*, 2016; Kotarba *et al.*, 2017, 2020).

To evaluate the feasibility of calcite purification, we simulated the TSR reaction using HCh software under conditions relevant to deep burial diagenetic environments (150–200 °C, 100 MPa). For the CaSO₄-CH₄-H₂O system, it comprised 0.5 mol CH₄ and 1 kg H₂O, with stepwise additions of anhydrite (0.01 mol/step) to emulate progressive sulfate availability. The simulations confirmed the redox reaction:



where anhydrite dissolution drives methane oxidation, yielding calcite and H₂S as dominant products (Fig. 1a, b). At 150 °C, the system produced 0.465 mol calcite with residual 0.0338 mol/kg CO₂ and 0.423 mol/kg H₂S (plus 0.0721 mol/kg HS[−]; Fig. 1c); at 200 °C, calcite precipitation increased to 0.472 mol, while CO₂ declined to 0.0277 mol/kg and H₂S rose to 0.437 mol/kg (Fig. 1d). H₂S concentration consistently exceeded CO₂ by an order of magnitude, underscoring the efficiency of CO₂ sequestration *via* calcite formation.

To address natural brine chemistry, we incorporated 7.6 wt. % CaCl₂ and tested pH conditions (neutral to slightly alkaline: ΔpH = 0–0.2) to account for TSR’s hydrogen demand (Truche *et al.*, 2014) and carbonate buffering (Machel, 2001) (Fig. 1e, f). Under these constraints, calcite precipitation further

1. State Key Laboratory of Critical Earth Material Cycling and Mineral Deposits, School of Earth Sciences and Engineering, Nanjing University, Nanjing, Jiangsu 210023, China
 2. Frontiers Science Center for Critical Earth Material Cycling, Nanjing University, Nanjing, Jiangsu 210023, China
 3. Institute of Geology, Chinese Academy of Geological Sciences, Beijing 100037, China
 4. State Key Laboratory of Palaeobiology and Stratigraphy, Nanjing Institute of Geology and Palaeontology, Chinese Academy of Sciences, Nanjing, Jiangsu 210008, China
 5. University of Chinese Academy of Sciences, Beijing 100049, China
 6. CAS Key Laboratory of Experimental Study under Deep-sea Extreme Conditions, Institute of Deep-sea Science and Engineering, Chinese Academy of Sciences, Sanya 572000, China
 7. Civil and Resource Engineering School, University of Science and Technology Beijing, Beijing 100083, China
- * Corresponding author (email: xlinwang@nju.edu.cn)

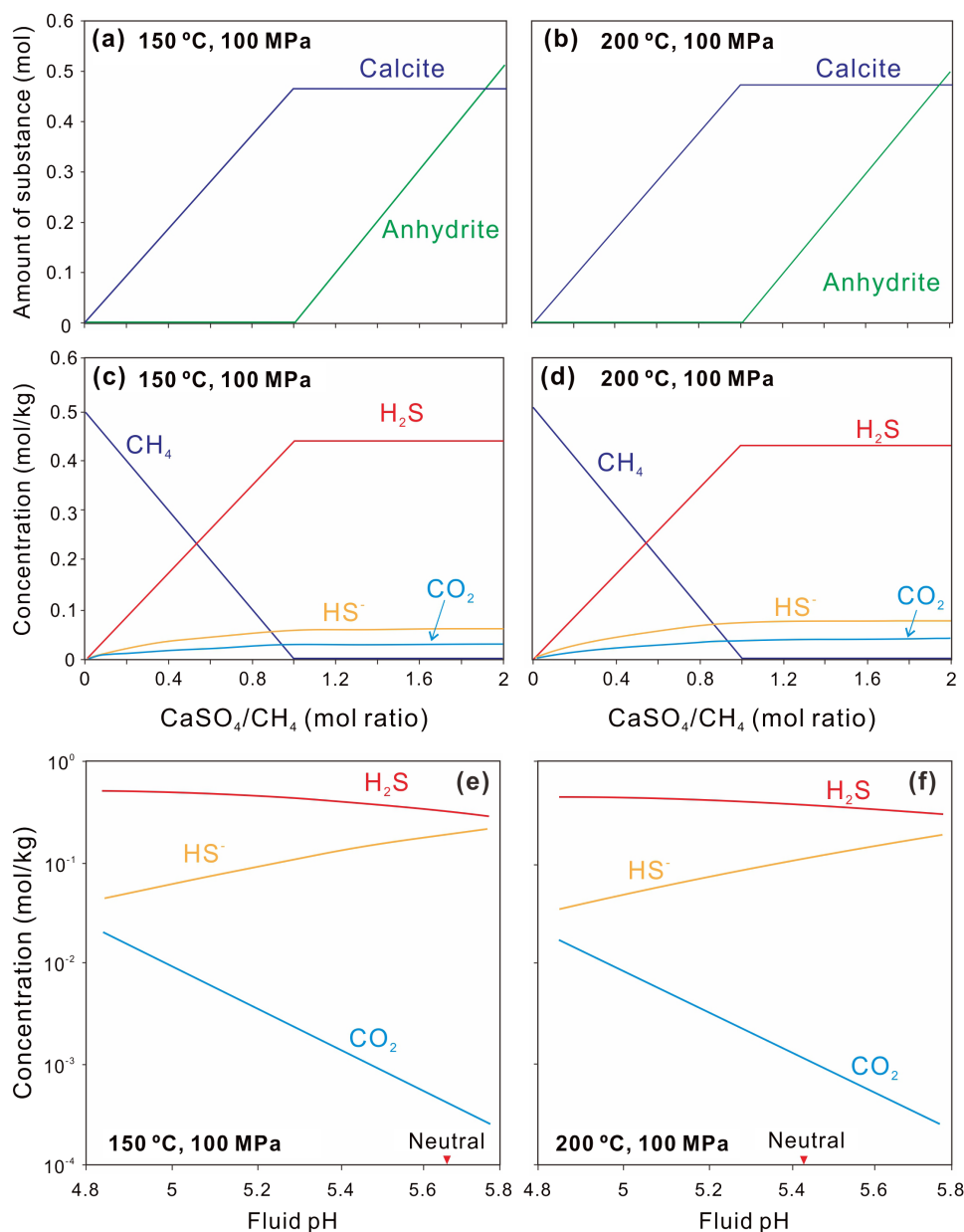


Figure 1 HCh simulation results of the TSR reactions. **(a–d)** $\text{CH}_4\text{-CaSO}_4\text{-H}_2\text{O}$ system; **(e–f)** $\text{CH}_4\text{-CaSO}_4\text{-CaCl}_2\text{-H}_2\text{O}$ system.

suppresses CO_2 concentrations to <0.001 mol/kg, while H_2S dominates the fluid phase (0.3–0.34 mol/kg). The resultant $\text{H}_2\text{S}/\text{CO}_2$ ratios ($\sim 660:1$) demonstrate that calcite acts as a CO_2 sink, enabling H_2S enrichment. If phase separation occurs, the volatile H_2S may partition into a gas phase, generating high purity H_2S fluids—a plausible mechanism for the observed H_2S predominance in TSR-altered reservoirs.

Cui (2025) proposed that the formation of large scale MVT Pb-Zn deposits requires external sulfur sources, which constitutes the essence of the fluid mixing model (e.g., Leach *et al.*, 2010). However, we emphasise that potential sulfur sources are not limited to microbial sulfate reduction, as TSR also represents a significant source of reduced sulfur. Regarding paragenetic sequence, while Cui (2025) suggested quartz cementation might predate calcite formation, our petrographic observations revealed no quartz rims lining the nodule. Instead, H_2S -rich fluid inclusions were exclusively found in quartz cements located within the nodules (Fig. 1b in Liu *et al.*,

2025), whereas only aqueous inclusions were identified in adjacent calcite cements, with no high density, H_2S -bearing inclusions detected. These observations lead us to conclude that the high density H_2S fluids were generated subsequent to calcite precipitation.

3. Complex Diagenetic and Burial Histories. Cui (2025) highlighted the complexity of the diagenetic history of the Neoproterozoic strata in South China. We have no disagreement with this understanding. We appreciate the recognition of our fluid inclusion observations. In comparison with calcite and dolomite, quartz has greater hardness and, generally, does not develop cleavage. As a result, fluid inclusions within quartz can often retain primary information even after undergoing complex diagenetic alteration. Nevertheless, further studies are still necessary to better reconstruct the diagenetic history.

Editor: Raúl Fonseca

Additional Information



© 2025 The Authors. This work is distributed under the Creative Commons Attribution Non-Commercial No-Derivatives 4.0

License, which permits unrestricted distribution provided the original author and source are credited. The material may not be adapted (remixed, transformed or built upon) or used for commercial purposes without written permission from the author. Additional information is available at <https://www.geochemicalperspectivesletters.org/copyright-and-permissions>.

Cite this letter as: Liu, Y., Wang, X., Song, Y., Shi, H., Chou, I-M., Wan, Q., Yu, C., Zhou, C. (2025) Reply to Comment on "Formation of abnormally high density H₂S fluid in sedimentary basins" by Liu *et al.*, 2025. *Geochem. Persp. Let.* 36, 20–22. <https://doi.org/10.7185/geochemlet.2529>

References

- BILDSTEIN, O., WORDEN, R.H., BROSE, E. (2001) Assessment of anhydrite dissolution as the rate-limiting step during thermochemical sulfate reduction. *Chemical Geology* 176, 173–189. [https://doi.org/10.1016/S0009-2541\(00\)00398-3](https://doi.org/10.1016/S0009-2541(00)00398-3)
- CUI, H. (2025) Comment on "Formation of abnormally high density H₂S fluid in sedimentary basins" by Liu *et al.*, 2025. *Geochemical Perspectives Letters* 36, 18–19. <https://doi.org/10.7185/geochemlet.2528>
- JENDEN, P.D., TITLEY, P.A., WORDEN, R.H. (2015) Enrichment of nitrogen and ¹³C of methane in natural gases from the Khuff Formation, Saudi Arabia, caused by thermochemical sulfate reduction. *Organic Geochemistry* 82, 54–68. <https://doi.org/10.1016/j.orggeochem.2015.02.008>
- JIANG, L., WORDEN, R.H., CAI, C.F. (2014) Thermochemical sulfate reduction and fluid evolution of the Lower Triassic Feixianguan Formation sour gas reservoirs, northeast Sichuan Basin, China. *AAPG Bulletin* 98, 947–973. <https://doi.org/10.1306/10171312220>
- KOTARBA, M.J., BILKIEWICZ, E., HALAS, S. (2017) Mechanisms of generation of hydrogen sulphide, carbon dioxide and hydrocarbon gases from selected petroleum fields of the Zechstein Main Dolomite carbonates of the western part of Polish Southern Permian Basin: Isotopic and geological approach. *Journal of Petroleum Science and Engineering* 157, 380–391. <https://doi.org/10.1016/j.petrol.2017.07.015>
- KOTARBA, M.J., BILKIEWICZ, E., KOSAKOWSKI, P. (2020) Origin of hydrocarbon and non-hydrocarbon (H₂S, CO₂ and N₂) components of natural gas accumulated in the Zechstein Main Dolomite carbonate reservoir of the western part of the Polish sector of the Southern Permian Basin. *Chemical Geology* 554, 119807. <https://doi.org/10.1016/j.chemgeo.2020.119807>
- KROUSE, H.R., VIAU, C.A., ELIUK, L.S., UEDA, A., HALAS, S. (1988) Chemical and isotopic evidence of thermochemical sulphate reduction by light hydrocarbon gases in deep carbonate reservoirs. *Nature* 333, 415–419. <https://doi.org/10.1038/333415a0>
- LEACH, D.L., BRADLEY, D.C., HUSTON, D., PISAREVSKY, S.A., TAYLOR, R.D., GARDOLL, S.J. (2010) Sediment-Hosted Lead-Zinc Deposits in Earth History. *Economic Geology* 105, 593–625. <https://doi.org/10.2113/gsecongeo.105.3.593>
- LI, P., HAO, F., GUO, X., ZOU, H., ZHU, Y., YU, X., WANG, G. (2016) Origin and distribution of hydrogen sulfide in the Yuanba gas field, Sichuan Basin, Southwest China. *Marine and Petroleum Geology* 75, 220–239. <https://doi.org/10.1016/j.marpetgeo.2016.04.021>
- LIU, Q., JIN, Z., WU, X., LIU, W., GAO, B., ZHANG, D., LI, J., HU, A. (2014) Origin and carbon isotope fractionation of CO₂ in marine sour gas reservoirs in the Eastern Sichuan Basin. *Organic Geochemistry* 74, 22–32. <https://doi.org/10.1016/j.orggeochem.2014.01.012>
- LIU, Y., WANG, X., SONG, Y., SHI, H., CHOU, I-M., ZHOU, C. (2025) Formation of abnormally high density H₂S fluid in sedimentary basins. *Geochemical Perspectives Letters* 34, 43–49. <https://doi.org/10.7185/geochemlet.2512>
- MACHEL, H.G. (2001) Bacterial and thermochemical sulfate reduction in diagenetic settings — old and new insights. *Sedimentary Geology* 140, 143–175. [https://doi.org/10.1016/S0037-0738\(00\)00176-7](https://doi.org/10.1016/S0037-0738(00)00176-7)
- TRUCHE, L., BAZARKINA, E.F., BARRÉ, G., THOMASSOT, E., BERGER, G., DUBESSY, J., ROBERT, P. (2014) The role of S₃[−] ion in thermochemical sulphate reduction: Geological and geochemical implications. *Earth and Planetary Science Letters* 396, 190–200. <https://doi.org/10.1016/j.epsl.2014.04.018>

WORDEN, R.H., SMALLEY, P.C., OXTOBY, N.H. (1996) The effects of thermochemical sulfate reduction upon formation water salinity and oxygen isotopes in carbonate gas reservoirs. *Geochimica et Cosmochimica Acta* 60, 3925–3931. [https://doi.org/10.1016/0016-7037\(96\)00216-5](https://doi.org/10.1016/0016-7037(96)00216-5)

ZHU, G., ZHANG, S., LIANG, Y., DAI, J., LI, J. (2005) Isotopic evidence of TSR origin for natural gas bearing high H₂S contents within the Feixianguan Formation of the Northeastern Sichuan Basin, southwestern China. *Science in China, Series D: Earth Sciences* 48, 1960–1971. <https://doi.org/10.1360/082004-147>