

Testing *ab initio* garnet-clinopyroxene Ca-isotope fractionation models with natural data

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Supplementary Information

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1. Materials

Selected eclogites are from the Garnet Ridge serpentinized ultramafic breccia (SUM) diatreme, and have been well-studied to understand their origin (Helmstaedt and Doig, 1975; Smith *et al.*, 2004), trace their metamorphic history (Smith and Zientek, 1979; Usui *et al.*, 2006) and the evolution of the Colorado Plateau (Hoover *et al.*, 2020; Roden *et al.*, 1990; Smith *et al.*, 2004). They are widely regarded as fragments of subducted oceanic crust, possibly having originated from the Farallon slab (e.g., Helmstaedt and Doig, 1975). Similar mineralogical descriptions of eclogites from the neighbouring Moses Rock SUM diatreme (Usui *et al.*, 2006) suggest these xenoliths are representative for the region.

Mineralogical description

Mineral modal abundances (averaged point counting and least squares regression) are shown in Table S-1. Samples N350 and N371 have been the subject of several previous studies, including detailed petrologic descriptions, in Smith and Zientek (1979), Roden *et al.* (1990), and Smith *et al.* (2004). Most pyroxenes and garnets have triple junctions between each other, and many samples are mineralogically zoned, with garnet- or lawsonite-rich zones, potentially along a previous metamorphic fabric or line of fluid flow. Many samples are crosscut by fine fractures. Garnets frequently occur in samples as both small euhedral grains (<1 mm) within a weak fabric defined by the pyroxene, as well as larger, euhedral grains (>2 mm) with spongy cores and heavily fractured rims. Garnets contain inclusions of euhedral pyroxene, and lawsonite, and pyroxenes contain inclusions of garnet. Pyroxene frequently appears to be

breaking down to lawsonite or zoisite. Pyroxene is typically <3 mm, with most larger grains clearly recrystallized along/near fractures. Rutile is present throughout all samples as a minor inclusion phase. Sample N350 contains minor, round, nodular pyrite grains up to 3mm across.

Table S-1 Lithology and mineral modal abundance for eclogitic xenoliths from Garnet Ridge, Navajo Volcanic Field.

Sample	Lithology	Mineralogy						
		cpx	grt	lawsonite	rutile	phengite	zoisite	pyrite
GR101	Eclogite	64	35	0	1			
GR105	Lawsonite eclogite	71	18	10	1			
GR110	Lawsonite eclogite	61	28	9	1	<1		
GR122	Eclogite	61	31	6	2			
GR130	Lawsonite-phengite eclogite	56	26	15	1	3		
GR182	Lawsonite eclogite	61	28	9	1	<1		
GR1j	Lawsonite eclogite	66	26	7	1			
N350-GR	Lawsonite-phengite eclogite	80	13	2.5	2	1.5		0.5
N371-GR	Eclogite	59	31	0	1	9	<1	<1

2. Analytical Methods

For grain mounts and stable isotope analysis, samples were crushed in a steel crusher, rinsed in dilute HCl and dried overnight in an oven. Optically clear and inclusion-free grain fragments of garnet and clinopyroxene were picked by hand. Because larger garnets often have spongy cores, this process likely preferentially sampled smaller garnet grains and inclusion-free rims of larger grains. Given the zoned nature of many larger garnet grains, this process also increases the likelihood that selected grains are in chemical and isotopic equilibrium with associated pyroxene grains.

Ca Isotope Ratios

Mineral separates were dissolved in HF + HNO₃ in an Anton-Paar Multiwave 7000 microwave. From the dissolved sample, a ~10 µg aliquot of Ca was mixed with ⁴²Ca-⁴⁸Ca double spike to correct for mass fractionation, and the mixture was dried down. The column chemistry to separate Ca followed the method of Huang *et al.* (2011), with the sample dissolved in 2.5M HCl and passed through a Teflon column packed with 250 µl of BioRad AG50W-X12 cation exchange resin. The sample solution was eluted in 1 ml of 2.5M HCl, and the eluate was run through the chemistry a second time to reduce contamination from K and Ti. The resulting solution was dried down. Samples were then analysed on a TE Triton TIMS, with Ca dissolved in 10% HNO₃ and 10 µg of Ca was dried onto a single Re filament. Accuracy and precision of measurements were ensured by analysing the secondary standards IAPSO seawater (expected 1.88 ± 0.04‰, measured 1.92 ± 0.09‰, n=7), JP-1 (expected 1.14 ± 0.10‰, measured 1.22 ± 0.12‰, n=6) and BHVO-2 basalt (expected 0.80 ± 0.05‰, measured 0.91 ± 0.10‰, n=10). Expected values compiled from the geological standards reference database GeoRem (https://georem.mpch-mainz.gwdg.de/sample_query.asp). Repeat standard analyses yielded a single analysis precision of 0.13 ‰ or better (2σ) for δ^{44/40}Ca_{SRM915a} (Brooker *et al.*, 2024). The precision was improved by repeated analysis of samples, lowering the 2 standard error to 0.1‰ or less. <https://georem.mpch-mainz.gwdg.de>

Major elements

Major element compositions were determined on garnet and clinopyroxene in two sessions. In the first, epoxy-mounted, polished, carbon-coated pyroxene and garnet separates were analysed using the JEOL JXA-8200 electron microprobe in the Jackson School of Geosciences at the University of Texas at Austin. All analyses were made using a 10 nA beam current with a 15 keV accelerating voltage and a focused beam. Compositionally appropriate secondary standards were analysed periodically throughout each session to monitor analytical quality and instrumental drift (Pyrope NMNH 143968 for garnets and Diopside NMNH 117733 for clinopyroxenes). For both sessions peak count

times were 20s and background 10s, and spots with anomalous totals were filtered out. Detection limits are typically ~0.1 wt.%. In the second session, a subset was analysed at Texas A&M University on a Cameca SXFive, using a 15 kV, 20 nA beam with a 3 μm spot size. This session was conducted mostly on polished, carbon-coated thin sections and was aimed at determining the heterogeneity of pyroxenes and garnet. In most samples the core and rim major element compositions are very similar, with flat Ca concentration profiles. The variation in GR105 pyroxene major element composition is significant, though, and the effect of this on temperature estimates is considered later. Spots were all chosen to avoid ablating inclusions.

Major element metadata available online through the Texas Data Repository:
<https://doi.org/10.18738/T8/P7XX0E>.

3. Major Element Results

Table S-2 Major element data for garnet, clinopyroxene, lawsonite and zoisite from Garnet Ridge eclogites, Navajo Volcanic Field.

Sample	SiO ₂	sd	TiO ₂	sd	Al ₂ O ₃	sd	Cr ₂ O ₃	sd	FeO ^a	sd	MnO	sd	MgO	sd	CaO	sd	Na ₂ O	sd
Garnet																		
GR1j	37.38	0.28	0.22	0.13	20.79	0.43	-	0.43	29.21	0.94	1.77	1.09	3.12	0.4	8.15	0.84	0.01	0.02
GR101	37.24	0.64	0.09	0.08	21.64	0.31	-	0.31	28.35	1.22	1.015	1.01	4.49	1.91	6.32	1.78	0.03	0.04
GR105	38.52	0.7	0.13	0.11	21.4	0.51	0.02	0.51	27.89	1.34	1	0.53	5.78	2.01	5.96	2.09	0	0.02
GR110	37.96	0.51	0.06	0.05	20.92	0.46	0.02	0.46	29.09	0.64	1	0.72	3.98	0.9	7.15	1.25	0.02	0.02
GR122	37.1	0.43	0.2	0.14	20.97	0.52	-	0.52	29.51	0.75	1.55	0.89	3.03	1.3	7.55	1.2	0.05	0.02
GR130	37.99	0.52	0.06	0.07	21.18	0.62	0.05	0.62	29.34	0.51	1.19	0.71	4.27	1.3	6.5	1.74	0.02	0.03
GR182	37.66	0.46	0.07	0.07	20.97	0.34	0.02	0.34	31.47	1.18	0.85	0.42	3.36	0.49	6.01	0.75	0.01	0.02
N350	37.53	-	-	-	21.22	-	0.05	-	30.93	-	1.11	-	4.19	-	5.98	-	0.01	-
N371	37.82	0.29	0.1	0.08	21.44	0.73	-	0.73	29.3	2.07	0.37	0.29	5.75	1.49	5.34	0.07	0.02	0.02
Pyroxene																		
GR1j	55.61	0.43	0.05	0.03	9.29	1	0.01	1	7.94	0.68	0	0.01	7.14	0.39	11.57	0.58	7.43	0.34
GR101	56.38	0.35	0.07	0.04	10.56	1.01	0.03	1.01	5.04	1.17	0.04	0.04	8.1	0.95	11.61	1.39	7.67	0.71
GR105	57.29	0.7	0.07	0.03	15.59	3	0.03	3	5.18	1.86	0.04	0.04	4.39	1.06	6.51	1.84	10.46	1.15
GR110	55.93	0.34	0.05	0.03	9.19	0.72	0.02	0.72	5.76	0.44	0.05	0.04	8.74	0.42	12.93	0.67	6.86	0.36
GR122	55.46	0.26	0.07	0.06	11.45	0.94	0.02	0.94	7.23	0.61	0.04	0.01	5.28	0.28	8.6	0.71	9.23	0.29
GR130	56.64	0.41	0.06	0.03	12.89	1.52	0.05	1.52	4.81	1.82	0.01	0.01	6.85	0.91	10.43	1.1	8.29	0.57
GR182	56.57	0.37	0.04	0.03	12.78	1.61	0.04	1.61	4.96	0.28	0.02	0.01	6.52	0.94	10.32	1.52	8.39	0.85
N350	57.83	-	-	-	16.16	-	0.03	-	3.86	-	-	-	4.94	-	6.78	-	11.19	-
N371	56.3	0.54	0.07	0.04	10.99	1.22	0.02	1.22	5.13	1.64	0.04	0.04	7.72	1.1	11.27	1.86	7.8	1.01

^a Total iron reported as FeO.

Analyses by electron microprobe. Major oxides reported in weight percent, and are averages of n samples.

Italicized data taken from personal communication D Smith.

Table S-2 continued.

Sample	K ₂ O	sd	NiO	sd	Total	Mg	Mineral composition	n	grains
Garnet									
GR1j	-	-	-	-	100.63	15	Alm ₆₁ Grs ₂₃ Pyr ₁₂ Sps ₄	5	3
GR101	-	-	-	-	99.3	23	Alm ₆₁ Pyr ₁₉ Grs ₁₈ Sps ₂	5	3
GR105	0.03	0.02	-	-	100.74	27	Alm ₅₉ Pyr ₂₂ Grs ₁₇ Sps ₂	24	5
GR110	0.02	0.01	-	-	100.21	20	Alm ₆₂ Grs ₂₀ Pyr ₁₆ Sps ₂	25	6
GR122	-	-	-	-	99.73	15	Alm ₆₂ Grs ₂₂ Pyr ₁₂ Sps ₄	9	4
GR130	0.01	0.01	-	-	100.61	21	Alm ₆₂ Grs ₁₈ Pyr ₁₇ Sps ₃	18	5
GR182	0.01	0.01	-	-	100.42	16	Alm ₆₈ Grs ₁₇ Pyr ₁₃ Sps ₂	19	5
<i>N350</i>	<i>0.01</i>	-	-	-	<i>101</i>	19	Alm ₆₄ Grs ₁₇ Pyr ₁₇ Sps ₂	-	-
N371	-	-	-	-	100.14	19	Alm ₆₀ Grs ₁₉ Pyr ₁₆ Sps ₄	3	3
Pyroxene									
GR1j	0	0.01	-	-	99.04	73	Jd ₄₃ Di ₄₆ Ac ₁₁	22	6
GR101	0	0.01	0.03	0.03	99.51	85	Jd ₄₇ Di ₄₆ Ac ₇	25	7
GR105	0	0.01	0.03	0.04	99.55	67	Jd ₇₀ Di ₂₆ Ac ₄	29	9
GR110	0.01	0.01	0.02	0.02	99.53	84	Di ₅₁ Jd ₄₀ Ac ₉	24	7
GR122	-	0	0	0.01	97.39	75	Jd ₅₃ Di ₃₄ Ac ₁₃	4	2
GR130	0	0.01	0.04	0.04	99.56	77	Jd ₅₆ Di ₄₁ Ac ₃	24	7
GR182	0	0.01	0.02	0.03	99.67	75	Jd ₅₆ Di ₄₀ Ac ₃	32	9
<i>N350</i>	<i>0.02</i>	-	-	-	<i>100.82</i>	<i>70</i>	<i>Jd₇₅Di₂₅Ac₀</i>	-	-
N371	0	0.01	0.03	0.03	99.33	82	Jd ₄₉ Di ₄₄ Ac ₇	24	7

^a Total iron reported as FeO.

Analyses by electron microprobe. Major oxides reported in weight percent, and are averages of n samples. Italicized data taken from personal communication D Smith.

4. $\delta^{44/40}\text{Ca}$ Data

$\delta^{44/40}\text{Ca}$ values for NVF are summarized in Table S-3, and literature sample $\delta^{44/40}\text{Ca}$ values, CaO wt.% and modal abundances for pyroxenes and garnet are presented in Table S-4.

Table S-3 Garnet Ridge eclogite sample stable calcium isotope data.

Sample	Lithology	$\delta^{44/40}\text{Ca}$ vs SRM915a (‰) $\pm$ 2SE		
		garnet (n=2)	omphacite (n=2)	$\Delta^{44/40}\text{Ca}_{\text{grt-cpx}}^1$
GR101	Eclogite	1.68 $\pm$ 0.04	0.76 $\pm$ 0.02	0.92 $\pm$ 0.10
GR105	Lawsonite eclogite	1.70 $\pm$ 0.09	0.85 $\pm$ 0.01	0.85 $\pm$ 0.10
GR110	Lawsonite eclogite	1.80 $\pm$ 0.07	0.69 $\pm$ 0.04	1.11 $\pm$ 0.10
GR122	Eclogite	1.75 $\pm$ 0.04	1.35 $\pm$ 0.10	0.40 $\pm$ 0.10
GR130	Lawsonite-phengite eclogite	1.98 $\pm$ 0.04	1.11 $\pm$ 0.09	0.87 $\pm$ 0.11
GR182	Lawsonite eclogite	1.97 $\pm$ 0.08	1.21 $\pm$ 0.04	0.76 $\pm$ 0.10
GR1j	Lawsonite eclogite	1.77 $\pm$ 0.04	1.06 $\pm$ 0.08	0.71 $\pm$ 0.10
N350-GR	Lawsonite-phengite eclogite	1.68 $\pm$ 0.02	0.96 $\pm$ 0.01	0.72 $\pm$ 0.10
N371-GR	Eclogite	1.54 $\pm$ 0.04	0.90 $\pm$ 0.04	0.64 $\pm$ 0.10

¹ 2SE of long-term standards IAPSO and BHVO-2 used (0.10‰ per analysis, 0.14‰ propagated error, see Section 8).

The range of $\delta^{44/40}\text{Ca}$ values for clinopyroxene and garnet in mantle eclogites are shown in Figure S-1, together with literature values for: xenolithic eclogites from Roberts Victor (Chen *et al.*, 2020), Bellsbank (Smart *et al.*, 2021) and Udachnaya Kimberlites (Kang *et al.*, 2019); Dabie eclogites and Sulu garnet lherzolites from the Dabie-Sulu Orogen, North China (Wang *et al.*, 2019); xenolith garnet lherzolites from Slave Craton (Brooker *et al.*, 2024); sheared garnet peridotites from Premier Kimberlite, South Africa (Tappe *et al.*, 2021); and garnet pyroxenites from Hannouba Basaltic Plateau (Dai *et al.*, 2020). Overall, the clinopyroxene $\delta^{44/40}\text{Ca}$ values of NVF eclogites overlap and extend to slightly higher values than previous work, whereas the garnet $\delta^{44/40}\text{Ca}$ values are all significantly higher than most previous data (Fig. S-1).

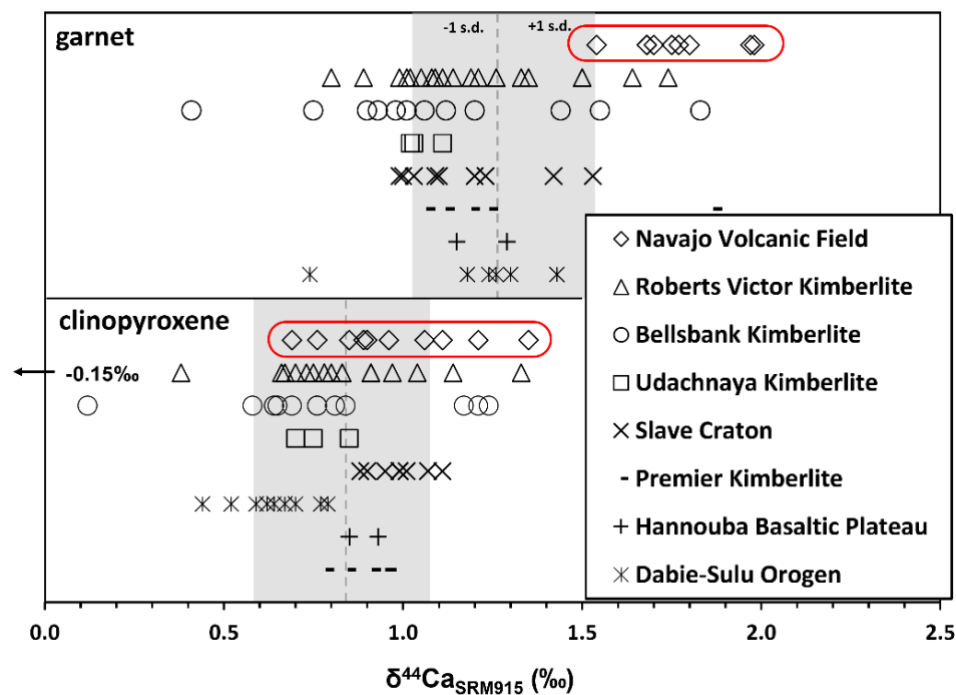


Figure S-1 Compilation of garnet and clinopyroxene calcium isotope composition ($\delta^{44/40}\text{Ca}$) of samples in this study (diamonds) compared to literature data for eclogites, garnet peridotites and pyroxenites (see text for references). The dashed lines and grey fields in represent the average and 1 σ , respectively.

Data for literature samples are included in Table S-4, where garnet and pyroxene modal abundances, CaO wt.% and $\delta^{44/40}\text{Ca}$ values are summarized.

Table S-4 Literature Ca isotope data, and garnet-pyroxene modal abundances and CaO contents.

Sample	Lithology	Modal abundance		CaO wt. %		$\delta^{44/40}\text{Ca}$ vs SRM915a (‰)			
		garnet	pyroxene	garnet	pyroxene	garnet	pyroxene	$\Delta^{44/40}\text{Ca}_{\text{grt-cpx}}$	2SE*
Roberts Victor Kimberlite, South Africa (Chen <i>et al.</i> , 2020)									
Type I									
RV07-02	Eclogite	28	73	14.36	4.25	1.19	0.83	0.36	0.08
RV07-03	Eclogite	62	38	14.84	5.20	1.09	0.91	0.18	0.12
RV07-09a	Eclogite	25	53	14.45	3.59	1.08	0.75	0.33	0.08
RV07-13	Eclogite	42	58	13.40	4.18	1.26	0.97	0.29	0.09
RV07-20	Eclogite	55	45	17.40	4.46	1.33	0.8	0.53	0.13
RVF-8	Eclogite	50	50	13.83	5.35	0.89	0.67	0.22	0.08
RV07-01	Eclogite	42	58	14.11	4.72	1.11	0.78	0.33	0.08
RV07-07	Eclogite	55	45	13.98	4.68	1.26	0.91	0.35	0.10
RV07-11	Eclogite	53	47	14.39	4.53	1.35	0.83	0.52	0.08
RV07-14	Eclogite	70	30	13.88	4.81	1.21	0.66	0.55	0.08
Type II									
RV07-12	Eclogite	48	52	18.04	6.79	1.5	0.7	0.8	0.12
RV73-12	Eclogite	45	55	18.63	4.26	0.8	-0.15	0.95	0.13
RV07-08	Eclogite	40	60	14.46	11.28	1.14	1.04	0.1	0.08
RV07-30	Eclogite	50	50	14.66	7.58	1.05	0.78	0.27	0.08
RV07-31	Eclogite	50	50	15.46	9.51	0.99	1.14	-0.15	0.08
RV07-33	Eclogite	50	50	15.05	10.35	1.01	0.91	0.1	0.12
RV07-34	Eclogite	50	50	14.33	13.58	1.02	1.33	-0.31	0.10
Bellsbank Kimberlite, South Africa (Smart <i>et al.</i> , 2021)									
DGE-1	Eclogite	55	45	16.63	3.86	0.9	0.76	0.14	0.13
DGE-2	Eclogite	55	45	16.41	12.16	1.01	0.76	0.25	0.07
DGE-3	Eclogite	55	45	16.69	3.86	0.93	0.58	0.35	0.10
DGE-4	Eclogite	55	45	17.12	10.47	1.55	1.21	0.34	0.10
DGE-8	Eclogite	55	45	15.98	10.67	1.2	0.84	0.36	0.10
DGO-5	Eclogite	55	45	15.62	10.81	1.12	0.81	0.31	0.10
DGO-6	Eclogite	55	45	15.26	9.01	0.98	0.69	0.29	0.10
DGO-7	Eclogite	55	45	16.95	11.10	1.83	1.24	0.59	0.07
DGO-8	Eclogite	55	45	17.44	5.07	0.41	0.12	0.29	0.10
DGO-11	Eclogite	55	45	15.08	11.17	1.06	0.64	0.42	0.10
DGO-14	Eclogite	55	45	13.41	11.22	1.44	1.17	0.27	0.10
DGO-15	Eclogite	55	45	11.48	15.28	0.75	0.65	0.1	0.10
Udachnaya Kimberlite, Russia (Kang <i>et al.</i> , 2019)									
U-400	Eclogite	55	45	14.20	7.52	1.03	0.85	0.18	0.11
U-401	Eclogite	55	45	20.56	12.75	1.02	0.7	0.32	0.11
U-402	Eclogite	55	45	16.88	4.72	1.11	0.75	0.36	0.11
Slave Craton, Canada (Brooker <i>et al.</i> 2024 & references therein)									
MX 158	Garnet lherzolite	10	5	19.543	5.09	1.42	0.90	0.52	0.14
MX5001	Garnet lherzolite	10	5	19.664	5.35	1.00	0.88	0.12	0.13
DDM-149	Garnet lherzolite	25	4	18.00	4.21	1.23	1.07	0.16	0.14
DDM-327	Garnet lherzolite	20	4	17.80	4.80	0.99	0.99	0.00	0.10
DDM-335	Garnet lherzolite	20	5	17.90	5.80	1.10	1.01	0.09	0.14
DDM-359	Garnet lherzolite	10	5	-	-	1.20	1.11	0.09	0.14
DDM-360	Garnet lherzolite	20	5	18.90	4.60	1.03	0.95	0.08	0.14
DDM-361	Garnet lherzolite	15	5	18.69	6.86	1.53	1.11	0.42	0.14
DDM-368	Garnet lherzolite	10	5	19.10	5.82	1.09	0.99	0.10	0.12

*2 σ of $\Delta^{44/40}\text{Ca}_{\text{grt-cpx}}$ using the standard deviation of internal standard used for study (e.g. BHVO-2, IAPSO, etc.), and propagating error using Eq. S-9 in Section 8.

Table S-4 continued.

Sample	Lithology	Modal abundance		CaO wt. %		$\delta^{44/40}\text{Ca}$ vs SRM915a (‰)			
		garnet	pyroxene	garnet	pyroxene	garnet	pyroxene	$\Delta^{44/40}\text{Ca}_{\text{grt-cpx}}$	2SE*
Premier Kimberlite, South Africa (Tappe <i>et al.</i> , 2021)									
CIM 15-003	Garnet peridotite	7	6	4.88	16.42	1.24	0.91	0.33	0.07
CIM 15-021	Garnet peridotite	7	4	4.75	16.18	1.06	0.96	0.11	0.07
CIM 15-022	Garnet peridotite	12	7	4.56	16.015	1.12	0.78	0.34	0.07
CIM 15-035	Garnet peridotite	15	5	6.085	18.775	1.19	0.84	0.35	0.07
CIM 15-040	Garnet peridotite	11	5	4.74	20.04	1.87	0.95	0.92	0.07
Hannouba Basaltic Plateau, China (Dai <i>et al.</i> , 2020)									
DMP441-PY	Garnet pyroxenite	39	41	19.6	4.21	1.15	0.93	0.22	0.11
DMP464-PV	Garnet pyroxenite	54	40	19.9	4.47	1.29	0.85	0.44	0.11
Dabie-Sulu Orogen, China (Wang <i>et al.</i> , 2019)									
BXL-1	Eclogite	54	33	14.43	7.73	0.74	0.59	0.15	0.03
BXL-4	Eclogite	52	34	14.05	8.27	1.26	0.64	0.62	0.03
BXL-6	Eclogite	60	30	13.58	8.21	1.43	0.62	0.81	0.04
HA-3	Eclogite	-	-	12.5	9.36	1.3	0.52	0.78	0.04
HA-4	Eclogite	-	-	12.92	10.5	1.24	0.44	0.8	0.04
SH-1	Eclogite	-	-	13.52	9.23	1.18	0.67	0.51	0.04
SH-5	Eclogite	-	-	14.31	8.4	1.18	0.7	0.48	0.04
ZMF-2	Garnet lherzolite	-	-	21.44	4.6	1.43	0.77	0.66	0.04
ZMF-3	Garnet lherzolite	-	-	22.46	4.65	1.43	0.79	0.64	0.05

*2 σ of $\Delta^{44/40}\text{Ca}_{\text{grt-cpx}}$ using the standard deviation of internal standard used for study (e.g. BHVO-2, IAPSO, etc.), and propagating error using Eq. S-9 in Section 8.

5. Estimation of Sample Equilibration Temperatures

Estimating temperatures

Because temperature estimates are important to compare natural data to the first principles models, the error and scatter produced by different methods for processing data must be mitigated. Across the limited dataset of Bellsbank, Roberts Victor, Premier, Hannouba and Dabie-Sulu samples there are two different garnet-pyroxene $\text{Fe}^{2+}\text{-Mg}^{2+}$ thermometers commonly used in previous studies, and three different Fe^{3+} corrections applied, whereas the Slave Craton temperatures reported in the literature are based on the Köhler and Brey (1990) clinopyroxene-orthopyroxene thermometer, which is not necessarily consistent with the garnet-clinopyroxene thermometer. To remove potential systematic biases in temperature estimates between different studies, all temperatures were recalculated using the same procedure outlined below. All Fe^{3+} corrections on clinopyroxene and garnet followed Droop (1987), and five thermometers were applied: those of Ellis and Green (1979), Krogh (1988), Powell (1985), Ravna (2000), and Sudholz *et al.* (2022). These thermometers are all based on the $\text{Fe}^{2+}\text{-Mg}^{2+}$ exchange between garnet and clinopyroxene, and subsequent ones build on previous work, with modified calibrations. These five thermometers were averaged to get a final temperature estimate.

In Figure S-2, it can be seen that the method applied in this reliably reproduces much of the literature temperature estimates within error (typically ≤ 50 °C; Krogh Ravna and Paquin, 2003). The standard deviation of temperature estimates from the five thermometers varied between 18 and 129 °C (average $1\sigma = 53$ °C). Erring on the side of caution, the temperature error allocated to each sample is the standard deviation from the averaging of thermometers or 50 °C, whichever is highest. The 50 °C threshold is because this is a typically reported error for Mg-Fe exchange thermometry (Krogh Ravna and Paquin, 2003).

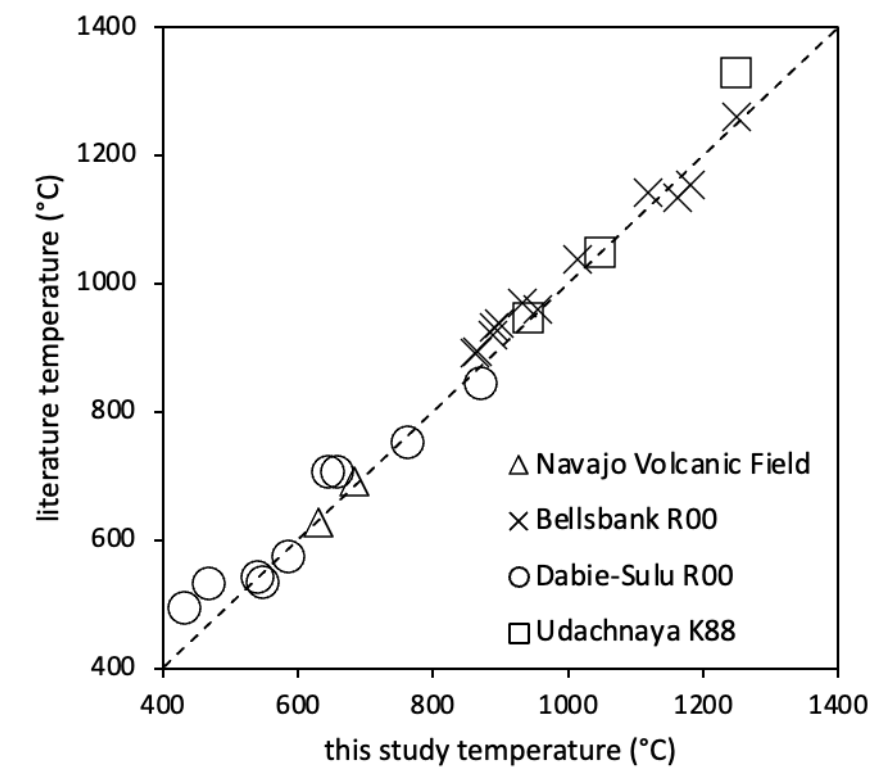


Figure S-2 Comparison of temperature estimates from literature and this study for eclogites from Dabie-Sulu (using Ravna, 2000), Bellsbank kimberlite (Ravna, 2000) Udachnaya (Krogh, 1988), and Navajo Volcanic Field (pseudosection work, Smith *et al.*, 2004).

Temperatures used for all samples are summarized in Table S-5 and literature temperatures are shown for reference. For NVF samples, temperatures were most consistent with published temperatures for GR1 and N350 (Smith *et al.*, 2004) when the pressure is set to 3.6 GPa.

Table S-5 Compilation of pressure and temperature estimates for samples from this study and literature. Literature pressure estimates are in the first column. The five Fe-Mg garnet-pyroxene thermometers used in this study are shown, and are averaged to yield the estimated temperature used in this study. Literature estimates of temperature are shown for comparison. Standard deviation is calculated from the five thermometers used to find an average temperature.

	P (GPa)	Temperature estimates in this study (°C)								Literature T (°C)
Sample	Lit	REE	Ellis&Green79	Powell85	Krogh88	Ravna00	Sudholz22	Average	1SD	
Navajo Volcanic Field (this study)										
GR101	3.6		550	523	477	514	564	526	30	
GR105	3.6		828	805	766	840	941	836	58	
GR110	3.6		547	520	482	507	543	520	34	
GR122	3.6		610	584	553	585	617	590	23	
GR130	3.6		625	599	557	600	656	607	33	
GR182	3.6		584	557	509	545	624	564	39	684
GR1j	3.6		638	613	585	617	636	618	19	
N350-GR	3.6		697	671	627	683	774	691	48	
N371-GR	3.6		569	542	502	548	577	548	26	
GR1	3.6		644	617	569	621	683	627	37	
										630

Table S-5 continued.

	P (GPa)	Temperature estimates in this study (°C)								Literature
Sample	Lit	REE	Ellis&Green79	Powell85	Krogh88	Ravna00	Sudholz22	Average	1SD	T (°C)
Bellsbank Kimberlite (Smart <i>et al.</i> , 2021)										
DGE-1	4.3	854	1153	1140	1076	1174	1234	1155	51	1182
DGE-2	5.9	877	1252	1244	1293	1351	1164	1261	61	1249
DGE-3	4.2	818	1136	1122	1056	1150	1213	1135	50	1163
DGE-4	2.9	953	948	933	940	911	869	920	28	889
DGE-8	2.9	964	919	903	906	889	848	893	24	862
DGO-5	3	1014	971	957	967	927	871	939	37	899
DGO-6	2	1042	927	912	909	851	872	894	28	865
DGO-7	4	855	1059	1046	1067	1060	960	1038	40	1015
DGO-8	2	859	986	971	919	907	1022	961	43	956
DGO-11	3.2	876	1003	990	1004	964	895	971	41	933
DGO-14	2.8	711	967	953	963	914	862	932	40	891
DGO-15	5.3	605	1170	1162	1170	1210	1003	1143	72	1118
Roberts Victor Kimberlite (Huang <i>et al.</i> , 2012)										
Type I										
HRV77bimin	6		1261	1250	1225	1403	1437	1315	87	1191
HRV77megaC	6		1165	1150	1167	1266	1213	1192	43	1126
RV07-02	4.5	829	993	972	906	993	1067	986	52	1060
RV07-03	5.7	991	1119	1101	1073	1200	1217	1142	57	1205
RV07-09a	4.8	765	1056	1036	959	1099	1166	1063	68	1110
RV07-09b	6		1067	1045	969	1159	1211	1090	86	1099
RV07-13	4.4	851	1008	987	922	1037	1106	1012	60	1040
RV07-20	4.5	874	993	972	911	999	1046	984	44	1060
RV07-41	6		1226	1212	1136	1336	1412	1264	98	1075
RVF-5	6		1004	980	914	1029	1062	998	50	1112
RVF-6	6		922	896	851	972	972	923	46	1060
RVF-7	6		1196	1181	1154	1348	1383	1252	94	1125
RVF-8	4.5	961	992	971	933	998	1039	987	35	1050
RV07-01	4.5	836	1144	1130	1090	1181	1231	1155	48	1060
RV07-07	4.6	871	1004	984	929	1004	1059	996	42	1075
RV07-11	4.4	874	1118	1103	1055	1147	1207	1126	50	1055
RV07-14	4.5	907	1013	992	941	1009	1062	1003	39	1070
RV07-16	6	1152	1170	1156	1184	1251	1171	1187	33	1170
RV07-18	5.9	807	1227	1214	1190	1331	1349	1262	65	1138
RV07-19	6	1078	1194	1180	1189	1291	1256	1222	44	1162
RV07-22	5.9		1192	1176	1139	1283	1316	1221	67	1131
RV07-29a	6		1188	1173	1159	1282	1274	1215	52	1141
RV07-29b	6		1089	1069	1045	1147	1138	1098	39	1139
RV06 BG-1	6		1253	1245	1296	1369	1224	1277	51	1267
RV07-05a	6		1311	1309	1267	1380	1091	1272	97	1232
RV07-05b	6		1253	1244	1295	1355	1212	1272	49	1245
RV07-23a	6		1197	1184	1219	1300	1213	1222	41	1189
RV07-23b	6		1187	1173	1177	1286	1260	1217	47	1140
RV07-25	6		1014	993	994	1064	1005	1014	26	1192
Type II										
RV07-12	3.5	1007	914	894	870	902	935	903	22	915
RV07-36	5.7		1030	1007	947	1075	1127	1037	61	1066
RV07-66	4.5		1002	982	945	969	975	974	18	897
RV73-12	3.5	909	852	827	753	791	883	821	46	995
RV07-39	6		976	950	865	1022	1101	983	78	1013
BD1186	6		1239	1229	1278	1370	1246	1273	52	1220
BD1191	6		1220	1209	1252	1355	1267	1261	52	1194
RV07-08	5.1	884	1150	1139	1173	1236	1130	1166	38	1150
RV07-30	6.6	1127	1161	1144	1162	1271	1228	1193	48	1290
RV07-31	5.5	1068	1113	1099	1125	1181	1104	1125	30	1190
RV07-33	6.6	1104	1177	1163	1200	1290	1177	1202	46	1290

RV07-34	6.9	879	1325	1319	1366	1493	1303	1361	69	1320
RV07-37	6.1	1146	1163	1148	1160	1276	1247	1199	52	1188

Table S-5 continued.

	P (GPa)	Temperature estimates in this study (°C)								Literature
Sample	Lit	REE	Ellis&Green79	Powell85	Krogh88	Ravna00	Sudholz22	Average	1SD	T (°C)
Udachnaya Kimberlite (Kang <i>et al.</i> , 2019)										
U-400	5		1033	1015	1017	1098	1092	1051	36	1047
U-401	5		977	961	965	992	843	948	53	940
U-402	5		1314	1309	1294	1394	1347	1331	36	1247
Slave Craton (Brooker <i>et al.</i> 2024 & references therein)										
MX 158	6.8		969	943	897	973	917	940	29	1200
MX5001	6.2		1146	1128	1105	1182	1102	1132	30	1261
DDM_149	5.7	942	1234	1221	1178	1289	1240	1232	36	1255
DDM_327	6.6	1132	1366	1359	1352	1505	1393	1395	57	1288
DDM_335	6.3		1273	1262	1265	1358	1237	1279	41	1332
DDM_360	5.8	955	1315	1307	1287	1405	1320	1327	41	1257
DDM_361	8	1653	1280	1265	1290	1411	1236	1297	60	1305
DDM_367	6.3	1378	1048	1027	1014	1065	765	984	111	1285
DDM_368	5.9	1214	1227	1214	1210	1286	919	1171	129	1247
Premier Kimberlite (Tappe <i>et al.</i> , 2021)										
CIM 15-003	5.2	1170	1171	1156	1124	1184	1127	1152	24	1299
CIM 15-021	5.5	1170	1215	1202	1171	1252	1190	1206	27	1302
CIM 15-022	5.3	1107	1235	1224	1190	1276	1222	1229	28	1303
CIM 15-035	4.8	1186	999	979	951	960	928	963	24	1088
CIM 15-040	2.8	866	910	889	823	795	856	854	42	809
Hannouba Basaltic Plateau, China (Dai <i>et al.</i> , 2020)										
DMP441-PY	1.51		1117	1111	1054	1000	1084	1073	43	1004
DMP464-PV	1.26		1420	1440	1435	1394	1358	1409	30	1025
Dabie-Sulu Orogen (Wang <i>et al.</i> , 2019)										
BXL-1	3		625	600	565	549	529	573	34	586
BXL-4	3		575	550	519	518	502	533	26	549
BXL-6	3		583	558	528	527	513	542	25	541
HA-3	1.7		574	552	526	481	524	531	31	468
HA-4	1.7		540	518	492	445	468	493	34	432
SH-1	2.8		743	721	705	698	657	705	28	658
SH-5	2.8		739	717	697	697	675	705	21	645
ZMF-2	4.9		908	883	821	871	738	844	60	872
ZMF-3	4.9		820	793	727	763	658	752	56	764

It is assumed that all NVF eclogites formed at roughly the same depth and this value is consistent with suggested literature pressures of ~3.3 GPa (Smith *et al.*, 2004). Therefore, 3.6 GPa is used for all. For most samples, our temperature estimates and the literature values are within error. An exception is the Roberts Victor eclogites, as it appears that no Fe²⁺-Fe³⁺ correction was made in determining the literature temperature (Gréau *et al.*, 2011). ‘Average’ temperatures in Table S-5 were used in plotting Figure 1.

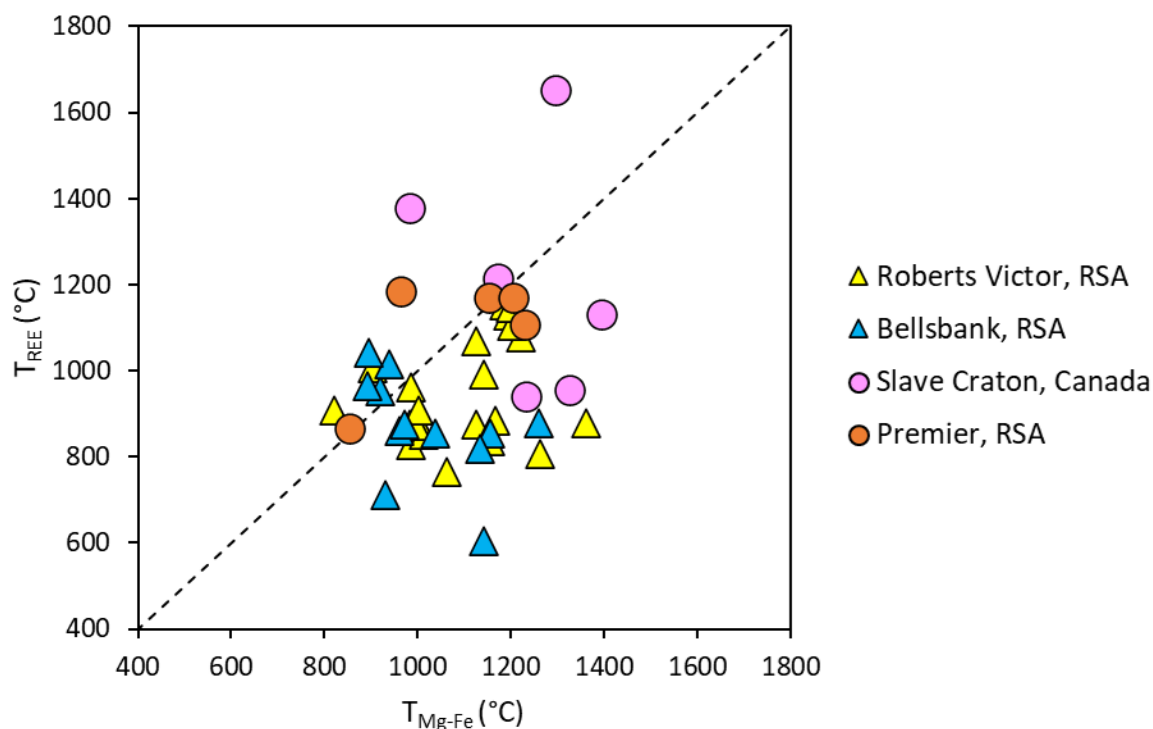


Figure S-3 Comparison of temperature estimates from Mg-Fe exchange thermometry and rare earth element in garnet and clinopyroxene thermobarometry of Sun and Liang (2015). Note that here, Mg-Fe thermometry typically yields higher temperature estimates.

The T_{REE} estimates are based on published literature data. Details such as where the REEs were measured (rims, cores, whole-grain averages), whether grains were zoned, number of duplicates, and whether garnet and clinopyroxene were in equilibrium are unknown, and therefore scatter in calculated T_{REE} estimates may reflect these uncertainties. Any apparent trends should be interpreted with caution, beyond noting that $T_{\text{Mg-Fe}}$ and T_{REE} values are often not aligned on a 1:1 line.

Temperature zonation in NVF eclogites

Because the grains in the NVF eclogites are zoned, it is important to ensure that the calculated temperatures are representative. Four samples were analysed for their mineral grain core and rim major element compositions. For each, three garnet and three clinopyroxene grains were measured and it was assumed that the rims/cores of the garnet were in equilibrium with the rims/cores of the clinopyroxene. The estimated temperatures are shown in Table S-6. The rims all record higher temperatures, showing that the growth of garnet and jadeite was on a prograde path. Although some of these temperature variations are large, it can be seen from Figure S-4 that the core and rim temperature estimates are mostly within the standard error expected from the thermometry (generally 50 °C), and that the zonation does not shift the predicted bulk temperature significantly. Hence, as the Ca isotope values are determined on bulk garnet and jadeite grains, the ‘bulk temperature’ is calculated from the average garnet and clinopyroxene major element content. For GR105, the core-rim temperature variation is large and the sample has clearly been heated significantly prior to emplacement. For this sample the average is still used, to keep the method consistent with temperature estimates from literature.

Table S-6 Summary of temperature estimates for core and rims of zones garnet-clinopyroxene pairs in NVF eclogites.

sample	Temperature (°C)			
	bulk	core	rim	rim - core
GR105	836	759	895	137
GR110	531	529	543	13
GR130	607	588	633	45
GR182	564	550	603	53

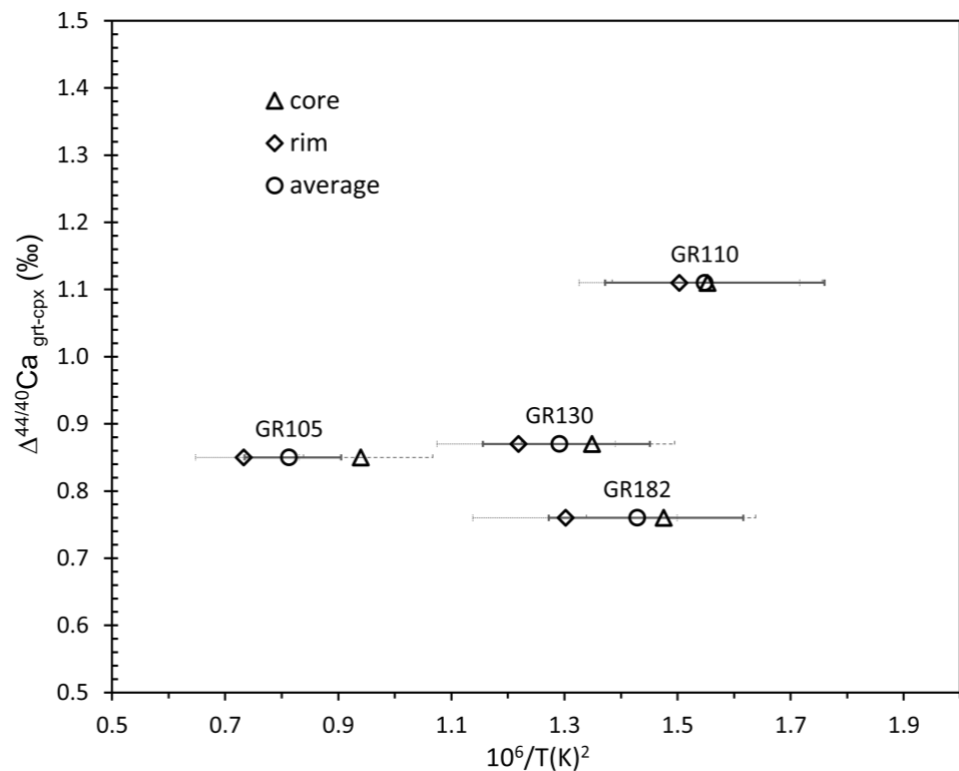


Figure S-4 Comparison of temperature estimates for Navajo Volcanic Field eclogites using: averaged garnet and clinopyroxene major element data to get average, whole-grain temperatures (circles), core-core garnet-clinopyroxene temperature estimates (triangles) and rim-rim temperature estimates (diamonds). Bold error bars are 1 standard error for the average grain temperature estimate, assuming a temperature estimate error of ± 50 °C.

6. Analysis of Fractionation Model and Residuals

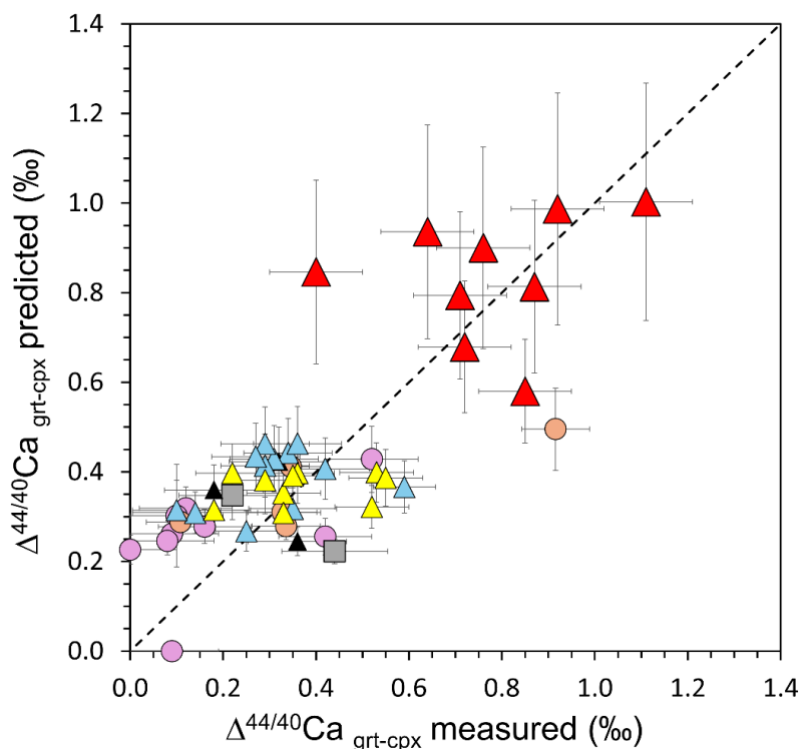


Figure S-5 $\Delta^{44/40}\text{Ca}_{\text{grt-cpx}}$ measured for natural samples versus the predicted $\Delta^{44/40}\text{Ca}_{\text{grt-cpx}}$ determined from Eq. 2. Error bars equivalent to 2SE, and symbols as for Figure 1.

Constraining the controls on natural $\Delta^{44/40}\text{Ca}_{\text{grt-cpx}}$ variability

To test the effect of garnet composition, garnet almandine, pyrope, and grossular proportions, and garnet CaO and Al_2O_3 wt.% were plotted against temperature-corrected residuals. To remove temperature effects, the measured $\Delta^{44/40}\text{Ca}_{\text{grt-cpx}}$ is divided by $10^6/T^2$. Then, the model value (Eq. 1) calculated at $10^6/T^2 = 1$ is subtracted from this adjusted $\Delta^{44/40}\text{Ca}_{\text{grt-cpx}}$ – this is the temperature-corrected residual. The temperature-corrected residual parameters used in Figure 2 are summarized in Table S-7. The dashed lines in Figure 2 represent the maximum expected compositional effects at 1000 K, assuming linear mixing. Expected compositional variation is typically less than the error on the residuals. Pressure variation based on Li *et al.* (2022) prediction of <0.05 ‰ at 1000 K from 1–10 GPa.

No statistically-significant correlations were seen, with the ‘best’ correlation being residual versus grossular content (p-value = 0.08, Fig. 2a). To test the effect of clinopyroxene composition, diopside content and CaO wt.% were plotted versus residuals. The lack of correlation is evident in Figure 2b where residual versus diopside content (p-value = 0.09) is shown. No jadeite effect is seen when jadeite content of pyroxenes is compared to residuals (p-value = 0.12, Fig. 2c). Pressure effects cannot be properly tested, since no accurate, independent geobarometer could be applied to all samples. However, where literature pressure estimates are plotted versus residuals (p-value = 0.31, Fig. 2d), no correlation is seen.

Table S-7 Data used for Figure 2, with measured $\Delta^{44/40}\text{Ca}_{\text{grt-cpx}}$ and $10^6/T^2$ used to calculate temperature-corrected residuals, measured $\Delta^{44/40}\text{Ca}_{\text{grt-cpx}}/10^6/T^2$ and temperature-corrected residuals. Almandine, grossular, pyrope, jadeite and diopside contents, and literature pressure estimates are included and plotted against residuals in Figure 2.

Sample	measured $\Delta^{44/40}\text{Ca}_{\text{grt-cpx}}$	$10^6/T^2$	measured $\Delta^{44/40}\text{Ca}_{\text{grt-cpx}} / 10^6/T^2$	Residual: (measured $\Delta^{44/40}\text{Ca}_{\text{grt-cpx}} / 10^6/T^2$) (predicted $\Delta^{44/40}\text{Ca}_{\text{grt-cpx}}$ at $10^6/T^2 = 1$)	almandine content	grossular content	pyrope content	jadeite content	diopside content	literature pressure estimate (GPa)
Navajo Volcanic Field, USA (this study)										
GR101	0.92	1.57	0.59	-0.04	61	18	19	47	46	3.6
GR105	0.85	0.92	0.92	0.29	59	17	22	70	26	3.6
GR110	1.11	1.59	0.70	0.07	62	20	16	40	51	3.6
GR122	0.4	1.34	0.30	-0.33	62	22	12	53	34	3.6
GR130	0.87	1.29	0.67	0.04	62	18	17	56	41	3.6
GR182	0.76	1.43	0.53	-0.10	68	17	13	56	40	3.6
GR1j	0.71	1.26	0.56	-0.07	61	23	12	43	46	3.6
N350-GR	0.72	1.08	0.67	0.04	64	17	17	75	25	3.6
N371-GR	0.64	1.49	0.43	-0.20	60	19	16	49	44	3.6
Roberts Victor Kimberlite (Huang <i>et al.</i>, 2012)										
RV07-02	0.36	0.63	0.57	-0.06	34	11	54	32	63	4.5
RV07-03	0.18	0.50	0.36	-0.27	35	14	50	28	65	5.7
RV07-09a	0.33	0.56	0.59	-0.04	34	9	55	26	67	4.8
RV07-13	0.29	0.61	0.48	-0.15	37	11	50	32	61	4.4
RV07-20	0.53	0.63	0.84	0.21	33	12	54	13	79	4.5
RVF-8	0.22	0.63	0.35	-0.28	34	14	51	35	59	4.5
RV07-01	0.33	0.49	0.67	0.04	29	12	57	40	60	4.5
RV07-07	0.35	0.62	0.56	-0.07	31	12	56	36	59	4.6
RV07-11	0.52	0.51	1.02	0.39	31	12	57	37	63	4.4
RV07-14	0.55	0.61	0.90	0.27	29	12	57	38	58	4.5
RV07-12	0.8	0.72	1.11	0.48	42	18	39	23	74	3.5
RV73-12	0.95	0.84	1.14	0.51	33	11	54	18	78	3.5
RV07-08	0.1	0.48	0.21	-0.42	39	30	30	42	58	5.1
RV07-30	0.27	0.47	0.58	-0.05	36	26	43	36	60	6.6
RV07-31	-0.15	0.51	-0.29	-0.92	36	26	38	35	62	5.5
RV07-33	0.1	0.46	0.22	-0.41	37	28	35	35	58	6.9
RV07-34	-0.31	0.37	-0.83	-1.46	35	37	28	42	58	6.9

Table S-7 continued.

Sample	measured $\Delta^{44}/40\text{Ca}_{\text{grt-cpx}}$	$10^6/T^2$	measured $\Delta^{44}/40\text{Ca}_{\text{grt-cpx}} / 10^6/T^2$	Residual: (measured $\Delta^{44}/40\text{Ca}_{\text{grt-cpx}} / 10^6/T^2$) (predicted $\Delta^{44}/40\text{Ca}_{\text{grt-cpx}} \text{ at } 10^6/T^2 = 1$)	almandine content	grossular content	pyrope content	jadeite content	diopside content	literature pressure estimate (GPa)
Bellsbank Kimberlite (Smart <i>et al.</i>, 2021)										
DGE-1	0.14	0.49	0.29	-0.34	27	10	63	28	72	4.3
DGE-2	0.25	0.43	0.59	-0.04	24	32	43	36	64	5.9
DGE-3	0.35	0.50	0.69	0.06	27	10	62	28	72	4.2
DGE-4	0.34	0.70	0.48	-0.15	31	27	41	34	66	2.9
DGE-8	0.36	0.74	0.49	-0.14	37	29	34	37	63	2.9
DGO-5	0.31	0.68	0.46	-0.17	26	28	45	40	60	3
DGO-6	0.29	0.73	0.40	-0.23	29	23	48	42	58	2
DGO-7	0.59	0.58	1.01	0.38	26	29	44	34	66	4
DGO-8	0.29	0.66	0.44	-0.19	31	13	55	31	69	2
DGO-11	0.42	0.65	0.65	0.02	24	29	47	42	58	3.2
DGO-14	0.27	0.69	0.39	-0.24	25	29	45	49	51	2.8
DGO-15	0.1	0.50	0.20	-0.43	22	40	38	56	44	5.3
Udachnaya Kimberlite (Kang <i>et al.</i>, 2019)										
U-400	0.18	0.57	0.32	-0.31	45	21	34	44	56	5
U-401	0.32	0.67	0.48	-0.15	32	34	33	18	82	5
U-402	0.36	0.39	0.93	0.30	18	12	69	15	85	5
Slave Craton (Brooker <i>et al.</i> 2024 & references therein)										
MX158	0.52	0.68	0.76	0.13	15	13	71	7	89	6.8
MX5001	0.12	0.51	0.24	-0.39	13	14	73	5	93	6.2
DDM_149	0.16	0.44	0.36	-0.27	13	11	76	14	84	5.7
DDM_327	0	0.36	0.00	-0.63	13	12	75	11	87	6.6
DDM_335	0.09	0.42	0.22	-0.41	11	15	73	10	87	6.3
DDM_360	0.08	0.39	0.20	-0.43	12	12	76	11	88	5.8
DDM_361	0.42	0.41	1.03	0.40	11	18	70	9	88	
DDM_368	0.1	0.48	0.21	-0.42	12	15	72	10	88	5.9
Hannouba Basaltic Plateau, China (Dai <i>et al.</i>, 2020)										
DMP441-PY	0.22	0.55	0.40	-0.23	14	11	74	15	85	1.51
DMP464-PV	0.44	0.35	1.25	0.62	10	11	78	16	84	1.26
Dabie-Sulu Orogen (Wang <i>et al.</i>, 2019)										
BXL-1	0.15	1.40	0.11	-0.52	37	21	42	40	56	3
BXL-4	0.62	1.54	0.40	-0.23	57	23	19	30	55	3
BXL-6	0.81	1.51	0.54	-0.09	57	23	19	32	53	3
HA-3	0.78	1.55	0.50	-0.13	65	27	8	40	50	1.7
HA-4	0.8	1.71	0.47	-0.16	63	30	5	40	52	1.7
SH-1	0.51	1.05	0.49	-0.14	54	26	19	42	54	2.8
SH-5	0.48	1.05	0.46	-0.17	58	24	17	35	58	2.8
ZMF-2	0.66	0.80	0.82	0.19	21	12	65	12	88	4.9
ZMF-3	0.64	0.95	0.67	0.04	21	12	66	5	95	4.9

Table S-7 continued.

Sample	measured $\Delta^{44/40}\text{Ca}_{\text{grt-cpx}}$	$10^6/T^2$	measured $\Delta^{44/40}\text{Ca}_{\text{grt-cpx}} / 106/T^2$	Residual: (measured $\Delta^{44/40}\text{Ca}_{\text{grt-cpx}} / 10^6/T^2$) (predicted $\Delta^{44/40}\text{Ca}_{\text{grt-cpx}}$ at $10^6/T^2 = 1$)	almandine content	grossular content	pyrope content	jadeite content	diopside content	literature pressure estimate (GPa)
Premier Kimberlite (Tappe <i>et al.</i> , 2021)										
CIM15-003	0.33	0.23	1.40	0.77	10	12	77	3	97	5.2
CIM15-021	0.11	0.27	0.40	-0.23	11	12	76	3	97	5.5
CIM15-022	0.34	0.25	1.35	0.72	12	11	76	3	97	5.3
CIM15-035	0.35	0.47	0.74	0.11	13	16	70	2	98	4.8
CIM15-040	0.92	0.79	1.16	0.53	14	12	74	0	100	2.8

7. Tests for Disequilibrium

To assess equilibrium, the garnet-clinopyroxene partitioning of Mg and Ti were compared to expected equilibrium values suggested by O'Reilly and Griffin (1995), and REE partitioning was compared with predictions from Sun and Liang (2015). The plots are included (Fig. S-6). Clinopyroxene and garnet Mg# are well correlated, and the average D_{Ti} is ~ 0.85 , close to the expected value of ~ 1 (O'Reilly and Griffin, 1995). Additionally, the REE partitioning equilibrium was tested using the spreadsheet of Sun and Liang (2015) and the literature samples are broadly consistent with chemical equilibrium (two typical examples shown for samples from Bellsbank and Slave Craton). For NVF samples, Sun and Liang (2015) criteria could not be used since the jadeite content is too high, but the clinopyroxene-garnet REE partitioning (Fig. S-7) is similar to that expected for equilibrium (Green *et al.*, 2000).

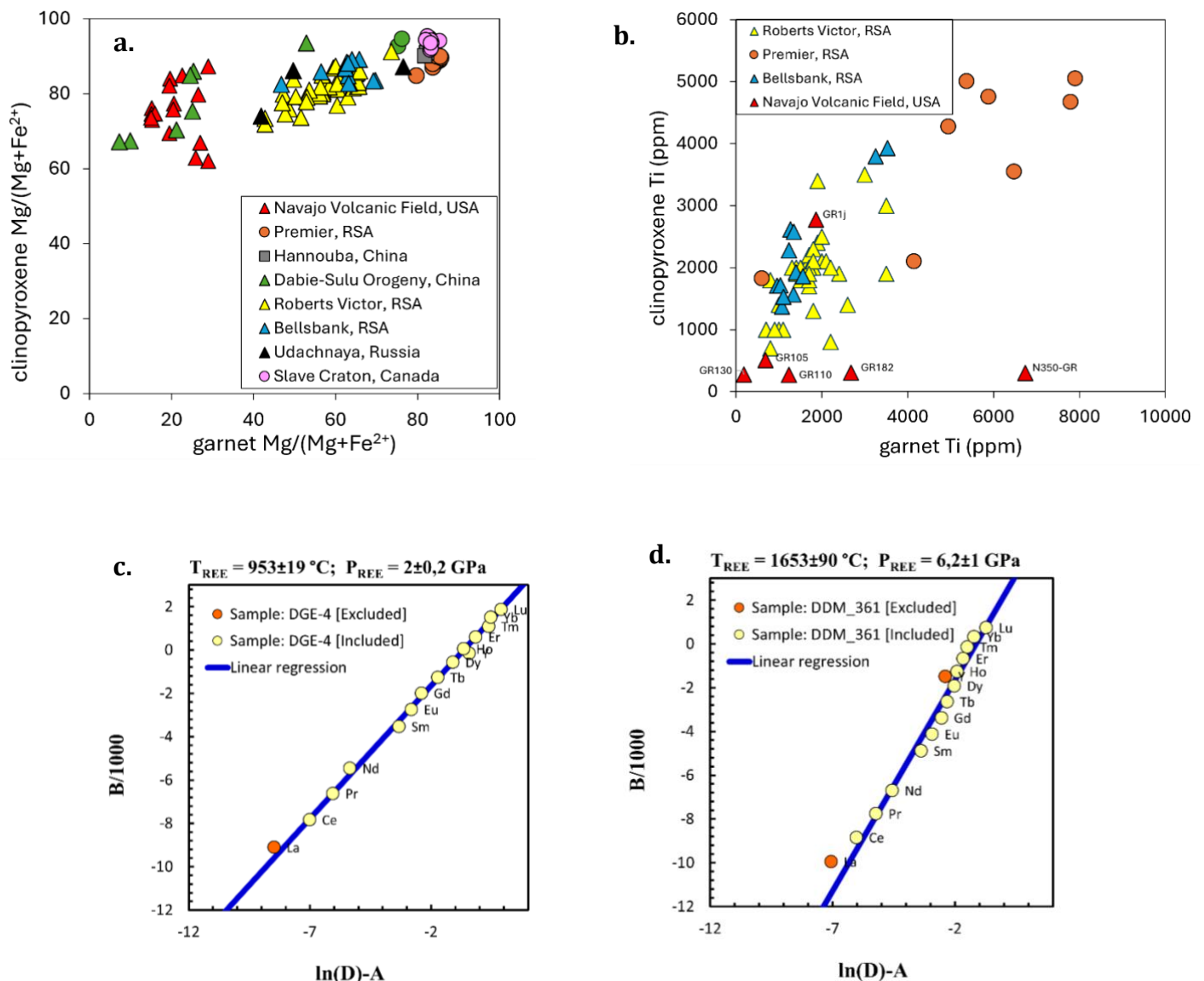


Figure S-6 Testing for disequilibrium in NVF sample and literature xenoliths with garnet versus clinopyroxene, a. Mg#, b. Ti, and REE partitioning in c. Bellsbank sample DGE-4 and d. Slave Craton sample DDM-361.

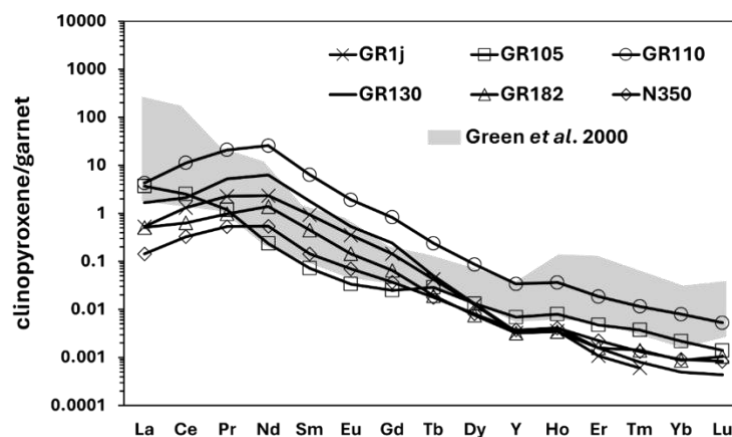


Figure S-7 Testing for disequilibrium in NVF sample, comparing clinopyroxene-garnet REE partitioning to experiments.

8. Regressions and Statistics

Below are the maximum likelihood linear regression results from Isoplot (Figure 1 data)

All data:

$$\Delta^{44/40}\text{Ca}_{\text{grt-cpx}} = 1.44 (\pm 0.15) \times 10^6/T^2 - 0.51 (\pm 0.09), \text{MSWD} = 7.5, n=65 \text{ (Eq. S-1)}$$

Excluding Roberts Victor Type II samples:

$$\Delta^{44/40}\text{Ca}_{\text{grt-cpx}} = 0.96 (\pm 0.11) \times 10^6/T^2 - 0.22 (\pm 0.07), \text{MSWD} = 6.4, n=58 \text{ (Eq. S-2)}$$

Excluding Dabie-Sulu and Roberts Victor Type II samples:

$$\Delta^{44/40}\text{Ca}_{\text{grt-cpx}} = 1.09 (\pm 0.14) \times 10^6/T^2 - 0.28 (\pm 0.08), \text{MSWD} = 5.7, n=49 \text{ (Eq. S-3)}$$

Below are the maximum likelihood linear regression results from Isoplot for Figure 1 with the intercept set at (0,0)

All data:

$$\Delta^{44/40}\text{Ca}_{\text{grt-cpx}} = 0.61 (\pm 0.02) \times 10^6/T^2, \text{MSWD} = 11, n=65 \text{ (Eq. S-4)}$$

Excluding Roberts Victor Type II samples:

$$\Delta^{44/40}\text{Ca}_{\text{grt-cpx}} = 0.62 (\pm 0.02) \times 10^6/T^2, \text{MSWD} = 7.2, n=58 \text{ (Eq. S-5)}$$

Excluding Dabie-Sulu and Roberts Victor Type II samples:

$$\Delta^{44/40}\text{Ca}_{\text{grt-cpx}} = 0.63 (\pm 0.03) \times 10^6/T^2, \text{MSWD} = 6.8, n=49 \text{ (Eq. S-6)}$$

Since equilibrium for Roberts Victor Type II eclogites is debated, they were excluded. Dabie-Sulu was then excluded since there is expected to be a $T_{\text{Mg-Fe}}$ and T_{Ca} mismatch. Eq. 2 is based on Eq. S-6 above and is the equation used or shown in each figure.

If the intercept is fixed at zero, the slope remains nearly unchanged regardless of whether the Dabie Sulu and Roberts Victor samples are excluded, indicating that consistency with theory does not depend on removing these data. The MSWD, though, improves significantly from 11 to 6.8, but is still quite far from the desired MSWD of 1. Hence, the Ca fractionation is not perfectly explained by the $1/T^2$ term (MSWD = 1), but rather there is still statistically significant scatter with MSWD of 6.8.

Statistics

Figure 1

x-axis ($10^6/T^2$): standard error (SE) is found by assuming that thermometry has an error of ± 50 °C or higher (see Table S-5), and applying equations below. 2 standard errors (2SE) are shown in the figure (simply 2 x SE).

Left-side standard error = $10^6/T^2 - 10^6/(T+50 \text{ °C})^2$ (Eq. S-7)

Right-side standard error = $10^6/(T-50 \text{ °C})^2 - 10^6/T^2$ (Eq. S-8)

y-axis (measured $\Delta^{44/40}\text{Ca}_{\text{grt-cpx}}$): standard error (2SE) is calculated using: $\text{SE}_{\Delta\text{Ca}} = \sqrt{(\text{SE}_{\delta\text{Ca-cpx}}^2 + \text{SE}_{\delta\text{Ca-grt}}^2)}$ (Eq. S-9), where SE_{δ} is the standard error of each mineral measurement. $\text{SE}_{\delta\text{Ca-cpx}}$ and $\text{SE}_{\delta\text{Ca-grt}}$ were assumed to each be equivalent to the average standard errors of standards run in each study (e.g. BHVO-2, IAPSO, or BCR-2). Hence, if IAPSO was analysed with samples and yielded a 2SE of 0.09, then $\text{SE}_{\delta\text{Ca-IAPSO}} = \text{SE}_{\delta\text{Ca-cpx}} = \text{SE}_{\delta\text{Ca-grt}} = 0.09 \text{ ‰}$. If the internal 2SE from a sample run is greater than the standard 2SE (here, 0.09‰), that larger 2SE is used for that mineral in Eq. S-10 above. Standard deviations (σ) are converted to SE with the relationship $\sigma = \text{standard error} \cdot \sqrt{n}$ (Eq. S-10), where necessary. 2 standard errors (2SE) are shown in the figure.

Figure 2

Expected error is presented as the normal distribution curve, centred at the mean of the residuals (-0.04) and $2\sigma = 0.14 \text{ ‰}$. The 2σ includes the propagated error from isotopic measurement $\Delta^{44/40}\text{Ca}_{\text{grt-cpx}}$, the error envelope from the linear regression Eq. 2 and $10^6/T^2$ estimate uncertainty: $\sigma_{\text{expected}} = \sqrt{(\sigma_{\text{measurement}}^2 + \sigma_{\text{temp estimate}}^2 + \sigma_{\text{error envelope}}^2)}$ (Eq. S-11).

Figure S-5

x-axis (measured $\Delta^{44/40}\text{Ca}_{\text{grt-cpx}}$): same as above for y-axis of Figure 1

y-axis (predicted $\Delta^{44/40}\text{Ca}_{\text{grt-cpx}}$): predicted $\Delta^{44/40}\text{Ca}_{\text{grt-cpx}}$ values were determined from the regression equation, hence their error is derived from $10^6/T^2$. 2SE for predicted $\Delta^{44/40}\text{Ca}_{\text{grt-cpx}}$ values are calculated as $0.63 \times 2\text{SE of } 10^6/T^2$.

Supplementary Information References

- Brooker, S.E., Barnes, J.D., Lassiter, J.C., Satkoski, A., Pearson, D.G. (2024) Tracking subduction-related metasomatism of the subcontinental lithospheric mantle using Ca-, O-, and H-isotopes. *Geochimica et Cosmochimica Acta* 385, 16–33. <https://doi.org/10.1016/j.gca.2024.09.036>
- Chen, C., Huang, J.-X., Foley, S.F., Wang, Z., Moynier, F., Liu, Y., Dai, W., Li, M. (2020) Compositional and pressure controls on calcium and magnesium isotope fractionation in magmatic systems. *Geochimica et Cosmochimica Acta* 290, 257–270. <https://doi.org/10.1016/j.gca.2020.09.006>
- Dai, W., Wang, Z., Liu, Y., Chen, C., Zong, K., Zhou, L., Zhang, G., Li, M., Moynier, F., Hu, Z. (2020) Calcium isotope compositions of mantle pyroxenites. *Geochimica et Cosmochimica Acta* 270, 144–159. <https://doi.org/10.1016/j.gca.2019.11.024>
- Droop, G.T.R. (1987) A general equation for estimating Fe^{3+} concentrations in ferromagnesian silicates and oxides from microprobe analyses, using stoichiometric criteria. *Mineralogical Magazine* 51, 431–435. <https://doi.org/10.1180/minmag.1987.051.361.10>
- Ellis, D.J., Green, D.H. (1979) An experimental study of the effect of Ca upon garnet-clinopyroxene Fe-Mg exchange equilibria. *Contributions to Mineralogy and Petrology* 71, 13–22. <https://doi.org/10.1007/BF00371878>
- Gréau, Y., Huang, J.-X., Griffin, W.L., Renac, C., Alard, O., O'Reilly, S.Y. (2011) Type I eclogites from Roberts Victor kimberlites: Products of extensive mantle metasomatism. *Geochimica et Cosmochimica Acta* 75, 6927–6954. <https://doi.org/10.1016/j.gca.2011.08.035>
- Green, T.H., Blundy, J.D., Adam, J., Yaxley, G.M. (2000) SIMS determination of trace element partition coefficients between garnet, clinopyroxene and hydrous basaltic liquids at 2–7.5 GPa and 1080–1200°C. *Lithos* 53, 165–187. [https://doi.org/10.1016/S0024-4937\(00\)00023-2](https://doi.org/10.1016/S0024-4937(00)00023-2)

- Helmstaedt, H., Doig, R. (1975) Eclogite nodules from kimberlite pipes of the Colorado plateau—samples of subducted Franciscan-type oceanic lithosphere. *Physics and Chemistry of the Earth* 9, 95–111. [https://doi.org/10.1016/0079-1946\(75\)90010-5](https://doi.org/10.1016/0079-1946(75)90010-5)
- Hoover, W.F., Page, F.Z., Schulze, D.J., Kitajima, K., Valley, J.W. (2020) Massive Fluid Influx beneath the Colorado Plateau (USA) Related to Slab Removal and Diatreme Emplacement: Evidence from Oxygen Isotope Zoning in Eclogite Xenoliths. *Journal of Petrology* 61, egaa102. <https://doi.org/10.1093/petrology/egaa102>
- Huang, J.-X., Gréau, Y., Griffin, W.L., O'Reilly, S.Y., Pearson, N.J. (2012) Multi-stage origin of Roberts Victor eclogites: Progressive metasomatism and its isotopic effects. *Lithos* 142–143, 161–181. <https://doi.org/10.1016/j.lithos.2012.03.002>
- Huang, S., Farkaš, J., Jacobsen, S.B. (2011) Stable calcium isotopic compositions of Hawaiian shield lavas: Evidence for recycling of ancient marine carbonates into the mantle. *Geochimica et Cosmochimica Acta* 75, 4987–4997. <https://doi.org/10.1016/j.gca.2011.06.010>
- Kang, J.-T., Ionov, D.A., Zhu, H.-L., Liu, F., Zhang, Z.-F., Liu, Z., Huang, F. (2019) Calcium isotope sources and fractionation during melt-rock interaction in the lithospheric mantle: Evidence from pyroxenites, wehrlites, and eclogites. *Chemical Geology* 524, 272–282. <https://doi.org/10.1016/j.chemgeo.2019.06.010>
- Köhler, T.P., Brey, G.P. (1990) Calcium exchange between olivine and clinopyroxene calibrated as a geothermobarometer for natural peridotites from 2 to 60 kb with applications. *Geochimica et Cosmochimica Acta*, 54, 2375–2388. [https://doi.org/10.1016/0016-7037\(90\)90226-B](https://doi.org/10.1016/0016-7037(90)90226-B)
- Krogh, E.J. (1988) The garnet-clinopyroxene Fe-Mg geothermometer — a reinterpretation of existing experimental data. *Contributions to Mineralogy and Petrology* 99, 44–48.
- Krogh Ravna, E.J., Paquin, J. (2003) Thermobarometric methodologies applicable to eclogites and garnet ultrabasites. In: Carswell, D.A., Compagnoni, R., Rolfo, F. (Eds.) *Ultrahigh Pressure Metamorphism*. Eötvös University Press, Budapest, 229–259. <https://doi.org/10.1180/EMU-notes.5.8>
- Li, Y., Wu, Z., Huang, S., Wang, W. (2022) Pressure and concentration effects on intermineral calcium isotope fractionation involving garnet. *Chemical Geology* 591, 120722. <https://doi.org/10.1016/j.chemgeo.2022.120722>
- O'Reilly, S.Y., Griffin, W.L. (1995) Trace-element partitioning between garnet and clinopyroxene in mantle-derived pyroxenites and eclogites: *P-T-X* controls. *Chemical Geology* 121, 105–130. [https://doi.org/10.1016/0009-2541\(94\)00147-Z](https://doi.org/10.1016/0009-2541(94)00147-Z)
- Powell, R. (1985) Regression diagnostics and robust regression in geothermometer/geobarometer calibration: the garnet-clinopyroxene geothermometer revisited. *Journal of Metamorphic Geology* 3, 231–243. <https://doi.org/10.1111/j.1525-1314.1985.tb00319.x>
- Ravna, K. (2000) The garnet–clinopyroxene Fe²⁺–Mg geothermometer: an updated calibration. *Journal of Metamorphic Geology* 18, 211–219. <https://doi.org/10.1046/j.1525-1314.2000.00247.x>
- Roden, M.F., Smith, D., Murthy, V.R. (1990) Chemical constraints on lithosphere composition and evolution beneath the Colorado Plateau. *Journal of Geophysical Research: Solid Earth* 95, 2811–2831. <https://doi.org/10.1029/JB095iB03p02811>
- Smart, K.A., Tappe, S., Woodland, A.B., Greyling, D.R., Harris, C., Gussone, N. (2021) Constraints on Archean crust recycling and the origin of mantle redox variability from the $\delta^{44/40}\text{Ca} - \delta^{18}\text{O} - f\text{O}_2$ signatures of cratonic eclogites. *Earth and Planetary Science Letters* 556, 116720. <https://doi.org/10.1016/j.epsl.2020.116720>
- Smith, D., Zientek, M. (1979) Mineral chemistry and zoning in eclogite inclusions from Colorado Plateau diatremes. *Contributions to Mineralogy and Petrology* 69, 119–131. <https://doi.org/10.1007/BF00371855>
- Smith, D., Connelly, J.N., Manser, K., Moser, D.E., Housh, T.B., McDowell, F.W., Mack, L.E. (2004) Evolution of Navajo eclogites and hydration of the mantle wedge below the Colorado Plateau, southwestern United States. *Geochemistry, Geophysics, Geosystems* 5, Q04005. <https://doi.org/10.1029/2003GC000675>
- Sudholz, Z.J., Green, D.H., Yaxley, G.M., Jaques, A.L. (2022) Mantle geothermometry: experimental evaluation and recalibration of Fe–Mg geothermometers for garnet-clinopyroxene and garnet-orthopyroxene in peridotite, pyroxenite and eclogite systems. *Contributions to Mineralogy and Petrology* 177, 77. <https://doi.org/10.1007/s00410-022-01944-3>

- Sun, C., Liang, Y. (2015) A REE-in-garnet-clinopyroxene thermobarometer for eclogites, granulites and garnet peridotites. *Chemical Geology* 393–394, 79–92. <https://doi.org/10.1016/j.chemgeo.2014.11.014>
- Tappe, S., Massuyeau, M., Smart, K.A., Woodland, A.B., Gussone, N., Milne, S., Stracke, A. (2021) Sheared Peridotite and Megacryst Formation Beneath the Kaapvaal Craton: a Snapshot of Tectonomagmatic Processes across the Lithosphere–Asthenosphere Transition. *Journal of Petrology* 62, egab046. <https://doi.org/10.1093/petrology/egab046>
- Usui, T., Nakamura, E., Helmstaedt, H. (2006) Petrology and Geochemistry of Eclogite Xenoliths from the Colorado Plateau: Implications for the Evolution of Subducted Oceanic Crust. *Journal of Petrology* 47, 929–964. <https://doi.org/10.1093/petrology/egi101>
- Wang, Y., He, Y., Wu, H., Zhu, C., Huang, S., Huang, J. (2019) Calcium isotope fractionation during crustal melting and magma differentiation: Granitoid and mineral-pair perspectives. *Geochimica et Cosmochimica Acta* 259, 37–52. <https://doi.org/10.1016/j.gca.2019.05.030>